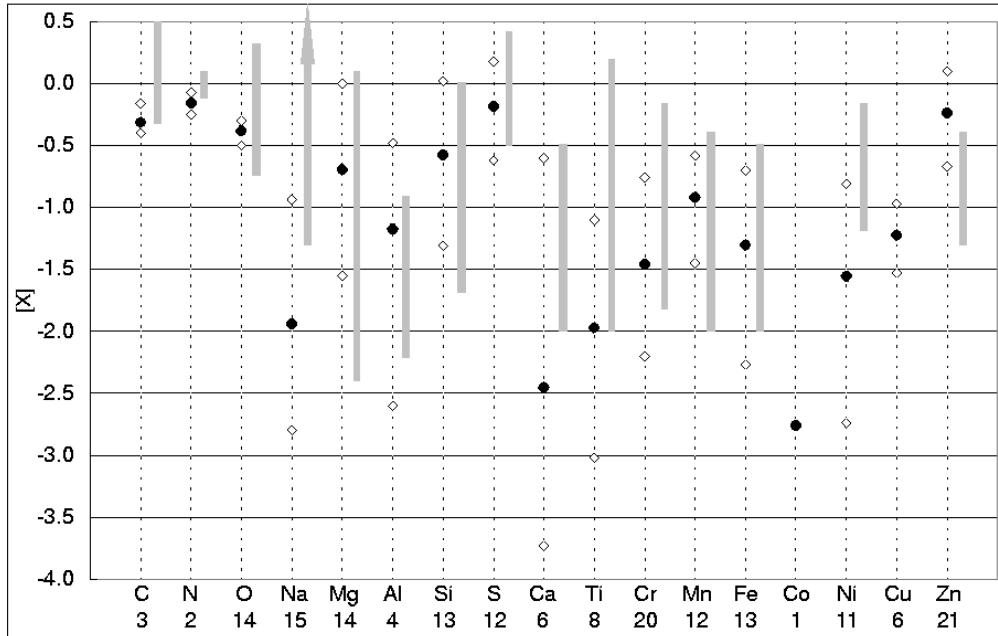


### Mean ISM abundances (black points)

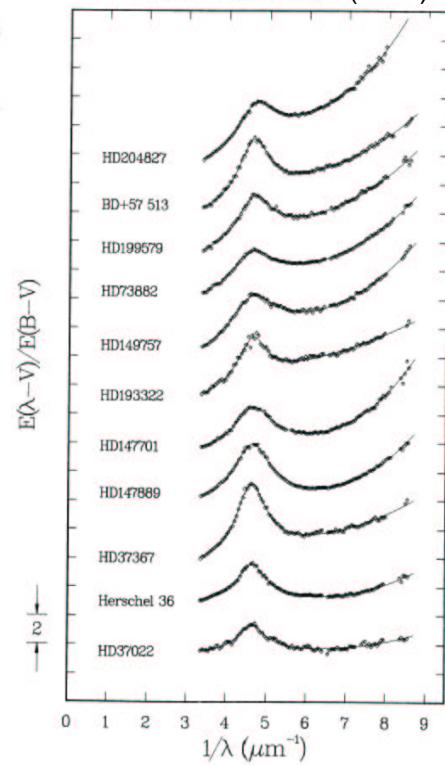


### Lodders (2003)

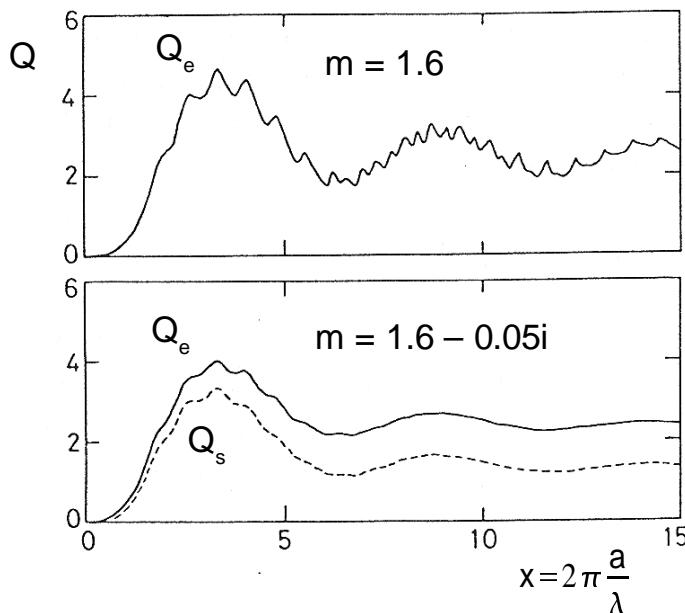
EQUILIBRIUM CONDENSATION TEMPERATURES FOR A SOLAR-SYSTEM COMPOSITION GAS

| Element<br>(1) | $T_C$<br>(K)<br>(2) | Initial Phase<br>{Dissolving Species}<br>(3)                           | 50% $T_C$<br>(K)<br>(4) | Major Phase(s) or Host(s)<br>(5)                         |
|----------------|---------------------|--|-------------------------|--|
| H              | 182                 | H <sub>2</sub> O ice   | ...                     | ...  |
| He             | <3                  | He ice   | ...                     | ...  |
| Li             |                     | {Li <sub>2</sub> SiO <sub>4</sub> , Li <sub>2</sub> SiO <sub>3</sub> } | 1142                    | Forsterite + enstatite                                   |
| Be             |                     | {BeCa <sub>2</sub> Si <sub>2</sub> O <sub>5</sub> }                    | 1452                    | Melilite   |
| B              |                     | {CaB <sub>2</sub> Si <sub>2</sub> O <sub>5</sub> }                     | 908                     | Feldspar   |
| C              | 78                  | CH <sub>4</sub> ·7H <sub>2</sub> O                                     | 40                      | CH <sub>4</sub> ·7H <sub>2</sub> O + CH <sub>4</sub> ice |
| N              | 131                 | NH <sub>3</sub> ·H <sub>2</sub> O                                      | 123                     | NH <sub>3</sub> ·H <sub>2</sub> O                        |
| O              | 182                 | Water ice <sup>a</sup>   | 180                     | rock + water ice   |
| F              | 739                 | Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> F                      | 734                     | F apatite  |
| Ne             | 9.3                 | Ne ice   | 9.1                     | Ne ice   |
| Na             |                     | {NaAlSi <sub>3</sub> O <sub>8</sub> }                                  | 958                     | Feldspar   |
| Mg             | 1397                | Spinel   |                         |  |
|                | 1354                | Forsterite <sup>b</sup>  | 1336                    | Forsterite   |
| Al             | 1677                | Al <sub>2</sub> O <sub>3</sub>   | 1653                    | Hibonite   |
| Si             | 1529                | Gehlenite  |                         |  |
|                | 1354                | Forsterite <sup>b</sup>  | 1310                    | Forsterite + enstatite                                   |
| P              | 1248                | Fe <sub>2</sub> P  | 1229                    | Schreibersite  |
| S              | 704                 | FeS  | 664                     | Troilite   |
| Cl             | 954                 | Na <sub>4</sub> [Al <sub>3</sub> Si <sub>3</sub> O <sub>12</sub> ]Cl   | 948                     | Sodalite   |
| Ar             | 48                  | Ar·6H <sub>2</sub> O   | 47                      | Ar·6H <sub>2</sub> O                                     |
| K              |                     | {KAlSi <sub>3</sub> O <sub>8</sub> }                                   | 1006                    | Feldspar   |
| Ca             | 1659                | CaAl <sub>12</sub> O <sub>19</sub>                                     | 1517                    | Hibonite + gehlenite                                     |
| Sc             |                     | {Sc <sub>2</sub> O <sub>3</sub> }                                      | 1659                    | Hibonite   |
| Ti             | 1593                | CaTiO <sub>3</sub>   | 1582                    | Titanate   |
| V              |                     | {VO, V <sub>2</sub> O <sub>3</sub> }                                   | 1429                    | Titanate   |
| Cr             |                     | {Cr}   | 1296                    | Fe alloy   |
| Mn             |                     | {Mn <sub>2</sub> SiO <sub>4</sub> , MnSiO <sub>3</sub> }               | 1158                    | Forsterite + enstatite                                   |
| Fe             | 1357 <sup>c</sup>   | Fe metal <sup>c</sup>  | 1334                    | Fe alloy   |
| Co             |                     | {Co}   | 1352                    | Fe alloy   |
| Ni             |                     | {Ni}   | 1353                    | Fe alloy   |
| Cu             |                     | {Cu}   | 1037                    | Fe alloy   |
| Zn             |                     | {Zn <sub>2</sub> SiO <sub>4</sub> , ZnSiO <sub>3</sub> }               | 726                     | Forsterite + enstatite                                   |

### FITZPATRICK AND MASSA (1988)



## Extinction efficiency for spherical grains from Mie theory



Refractive index:

$$m = m_R + m_I i$$

m<sub>R</sub> ... Scatteringm<sub>I</sub> ... Absorption

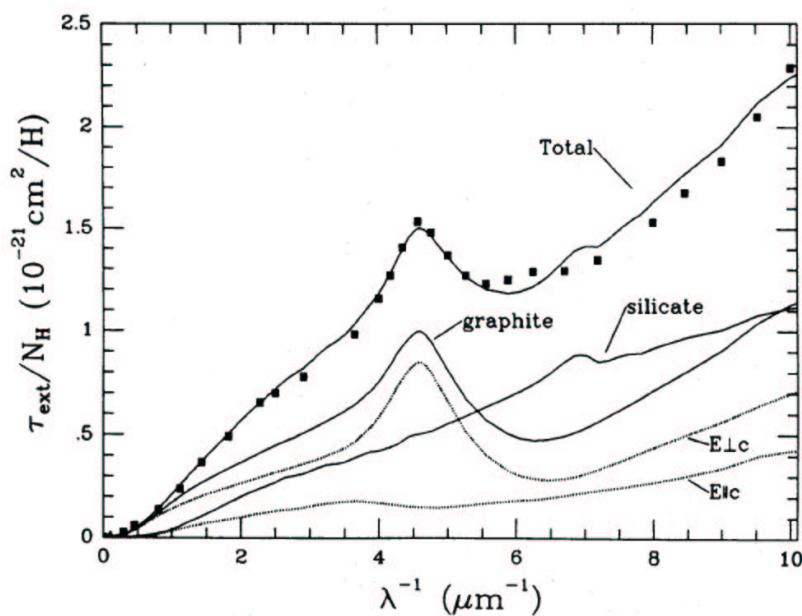
$$Q_e = Q_a + Q_s$$

Extinction cross-section

$$\sigma_e = \pi a^2 Q_e$$

a ... grain radius

λ ... wavelength

Whittet D.C.B, 1992, *Dust in the Galactic Environment*Average observed extinction  
and dust model of Draine & Lee (1984)

Size distribution:

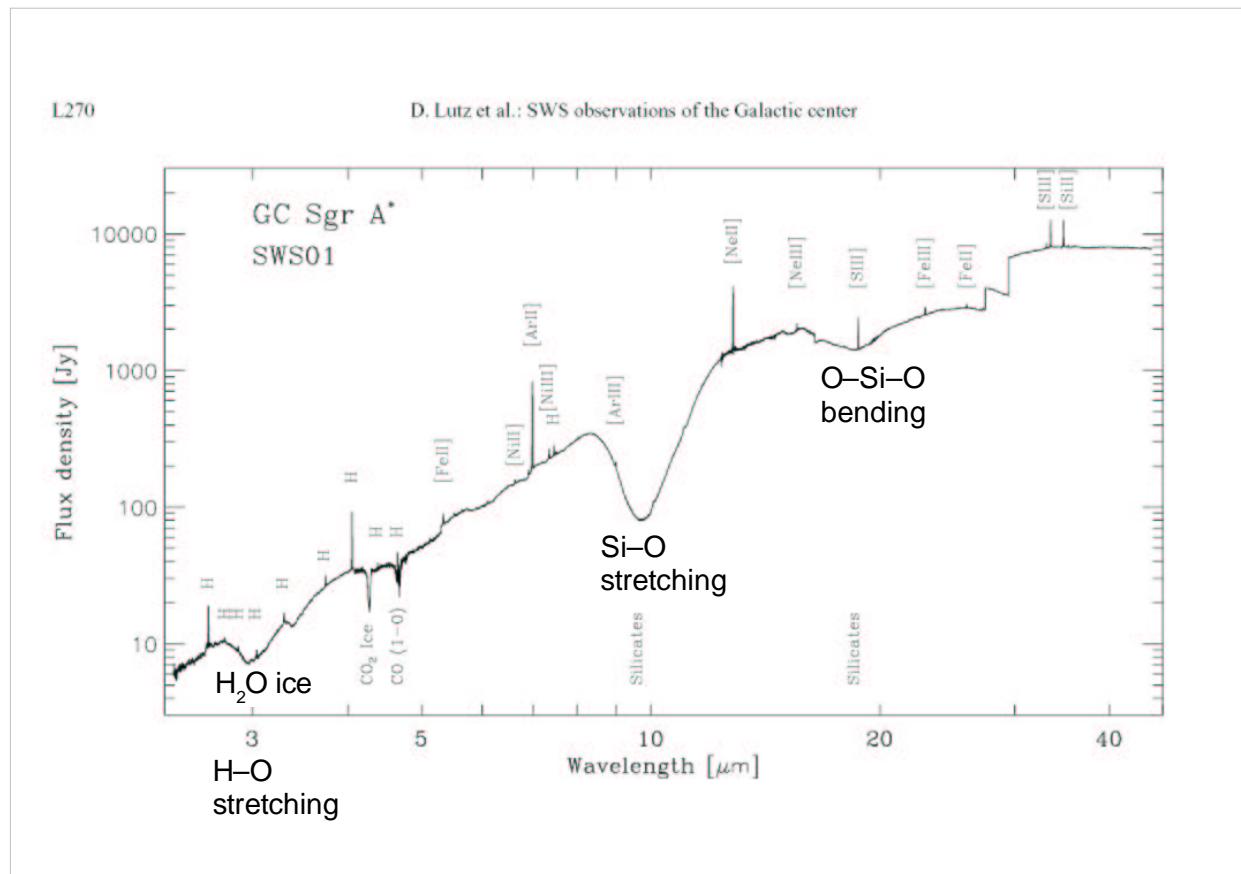
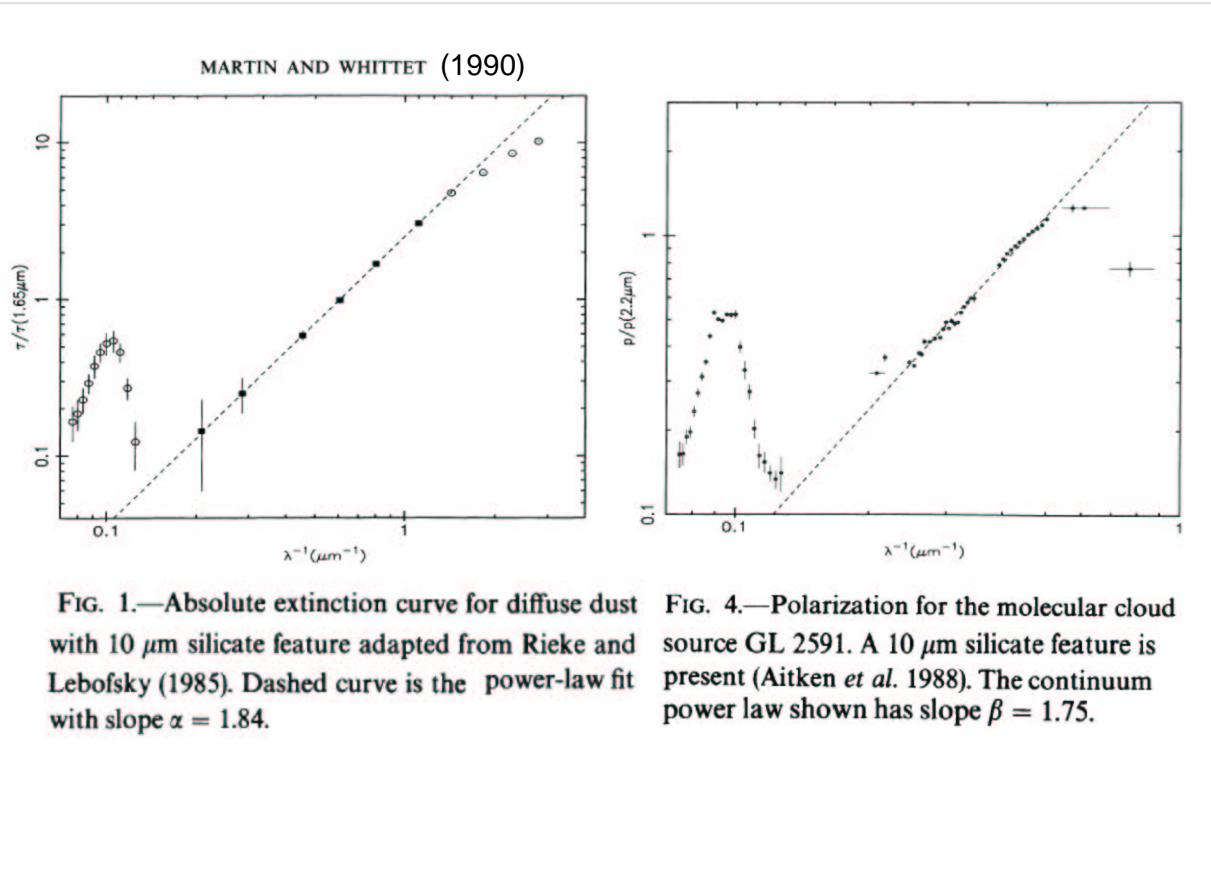
$$dn_i = A_i n_H a^{-3.5} da$$

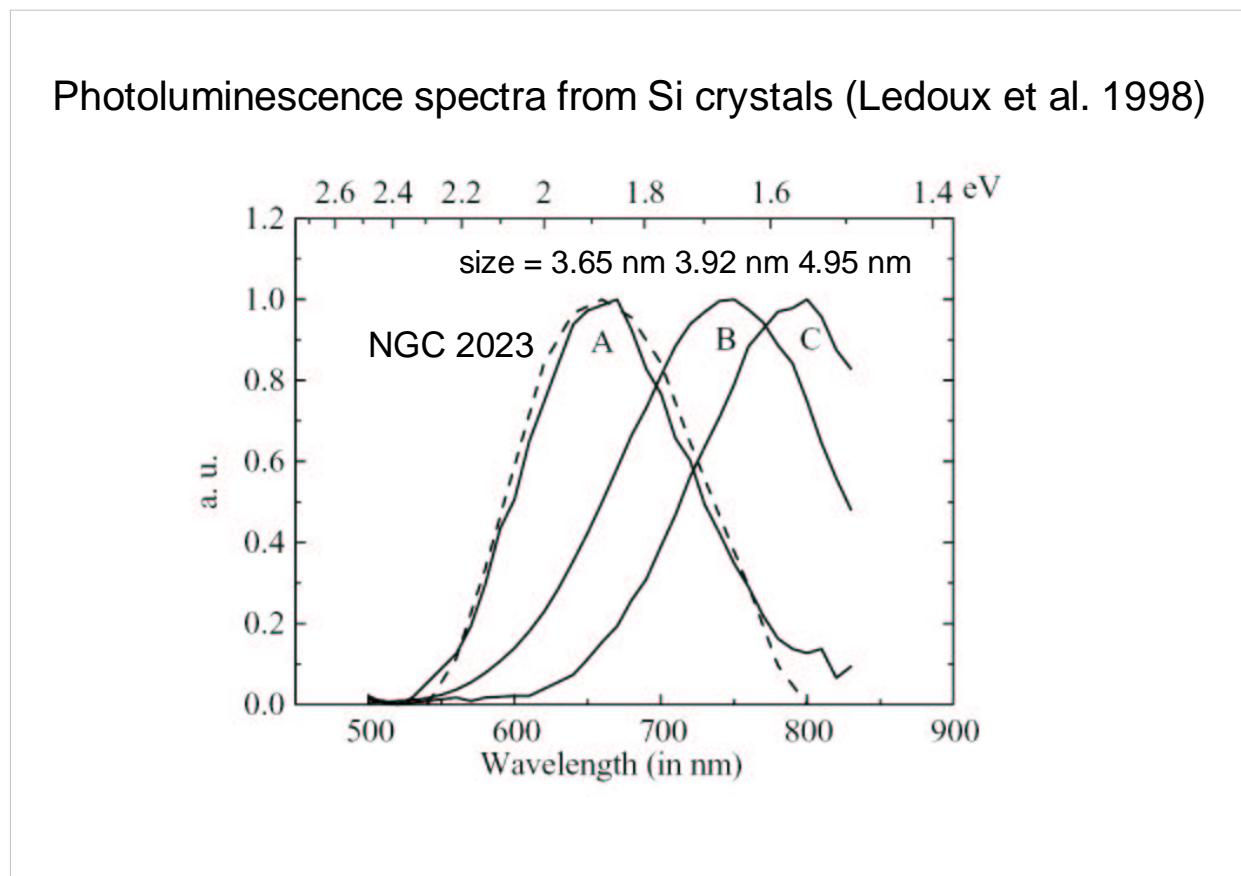
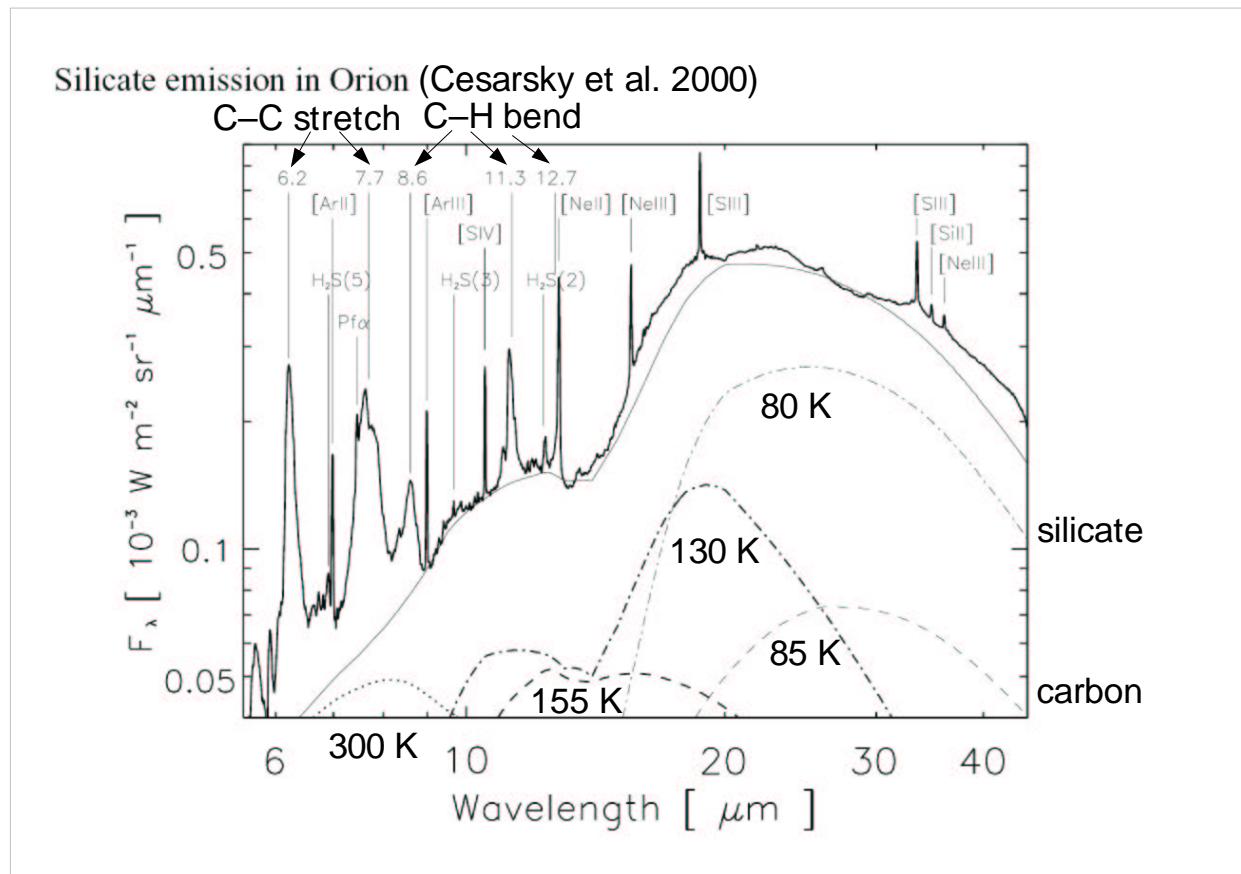
$$[a, a + da]$$

$$a_{\min} < a < a_{\max}$$

$$a_{\min} \approx 0.005 \mu\text{m}$$

$$a_{\max} \approx 0.25 \mu\text{m}$$





Witt et al. (1998)

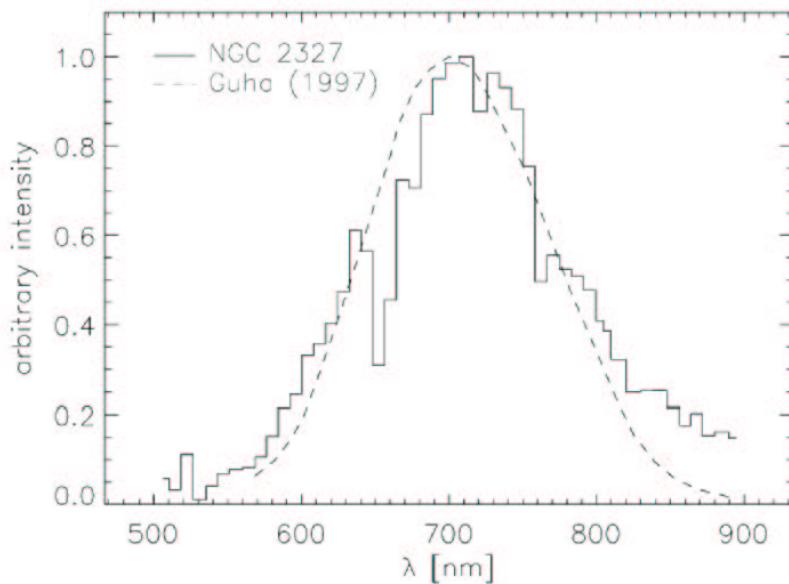
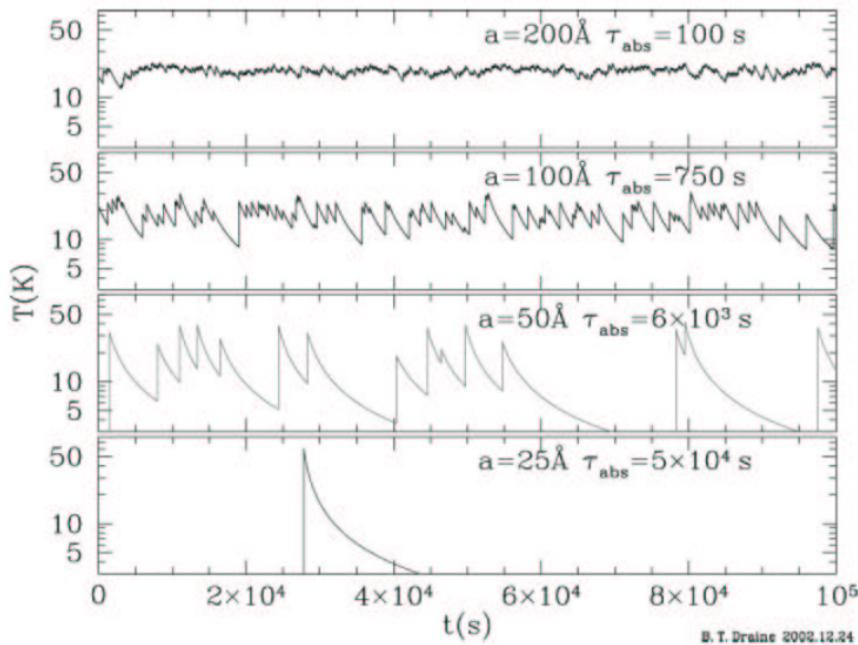


FIG. 1.—The observed ERE spectrum of NGC 2327 (Witt 1988) is plotted along with the photoluminescence of film *B* at room temperature from Guha (1997). Film *B* has a porosity of 70% that corresponds to a particle size of 4 nm.

#### Time dependence of grain temperature (Draine 2003)



**Figure 13** A day in the life of four carbonaceous grains, heated by the local interstellar radiation field.  $\tau_{\text{abs}}$  is the mean time between photon absorptions